

## Synthesis and Characterization of 1,4-Dichlorospiropentadiene

Rajesh K. Saini, Vladislav A. Litosh, Andrew D. Daniels, and W. E. Billups\*

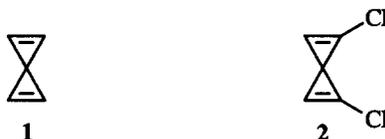
Department of Chemistry, Rice University, Houston, Texas 77005

Received 22 March 1999; revised 8 June 1999; accepted 14 June 1999

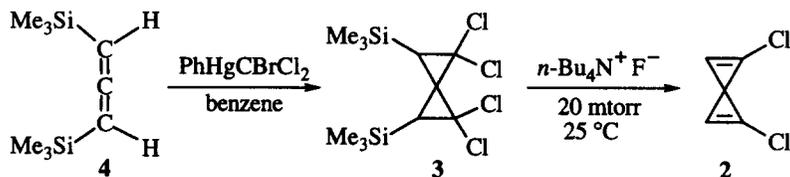
**Abstract:** 1,4-Dichlorospiropentadiene was prepared by vacuum gas-solid elimination of compound **3** over solid  $(n\text{-Bu})_4\text{N}^+\text{F}^-$  and characterized by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy at  $-103\text{ }^\circ\text{C}$ .

© 1999 Elsevier Science Ltd. All rights reserved.

The possibility that double bonds arranged perpendicularly in space might interact by conjugation has generated considerable interest in molecules with geometry suitable for this interaction.<sup>1,2</sup> Spiropentadiene **1** is of interest in this regard and although the parent hydrocarbon has been reported recently,<sup>3</sup> simple derivatives of this ring system have not been reported. Recent theoretical calculations lead to a standard heat of formation of 157.4 kcal/mole for **1**.<sup>4</sup> This is more than twice the experimental heat of formation of 66.2 kcal/mole for cyclopropene. We present here the synthesis and some of the properties of 1,4-dichlorospiropentadiene **2**.



The starting material, compound **3**,<sup>5</sup> was prepared in 75% yield by refluxing a solution of 1,3-trimethylsilyllallene **4** and  $\text{PhHgCBrCl}_2$ <sup>6</sup> in benzene for 48 hours. Elimination of trimethylsilylchloride from **3** to yield **2** using solid  $n\text{-Bu}_4\text{N}^+\text{F}^-$  adsorbed on glass helices as described previously for **1** could be effected



*in vacuo* at  $25\text{ }^\circ\text{C}$ . A  $^1\text{H}$  NMR (tetrahydrofuran-*d*8) signal at  $\delta$  7.72 could be observed for several minutes at  $-103\text{ }^\circ\text{C}$ . However, this signal disappeared rapidly as the temperature was raised.  $^{13}\text{C}$  NMR signals at 54.55 (51.26), 113.84 (112.2), and 123.62 (122.0) ppm are in agreement with chemical shifts calculated (shown in

parentheses) using MP2/6-311G(d,p) basis set at the MP2/6-311G(d,p) geometry.<sup>7</sup> Optimized bond lengths for **2** calculated at the same level of theory are presented in Figure 1.

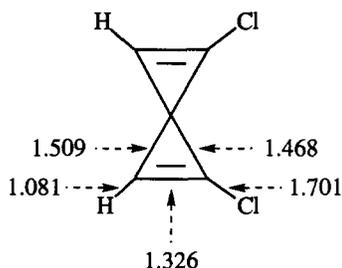
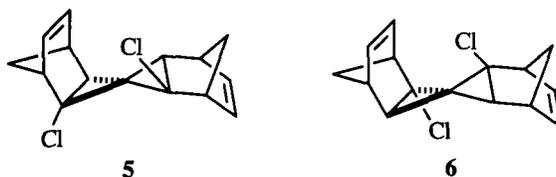


Figure 1. MP2/6-311G(d,p) geometry optimized bond lengths (Å) for **2**.

A mixture of products was formed when the spirene was condensed onto a cold surface coated with cyclopentadiene.<sup>3</sup> Two adducts in which both double bonds of the spirene had reacted (HRMS) were isolated in 8% combined yield by preparative TLC. Although assignment of stereochemistry was not possible, each compound exhibited a simple <sup>13</sup>C NMR spectrum consisting of eight lines. Since cyclopropenes normally yield *endo* addition products with cyclopentadiene, the <sup>13</sup>C NMR spectra can be rationalized in terms of compounds **5** and **6**. Structural assignments by x-ray crystallography are planned.



#### Acknowledgment

We gratefully acknowledge financial support from the National Science Foundation (CHE-9112530) and the Robert A. Welch Foundation.

#### References and Notes

1. Liebman, J. F.; Greenberg, A. *Chem. Rev.* **1976**, *76*, 311.
2. Review: Dürr, H.; Gleiter, R. *Angew. Chem. Int. Ed. Engl.* **1978**, *17*, 559.
3. Billups, W. E.; Haley, M. M. *J. Am. Chem. Soc.* **1991**, *113*, 5084.
4. Shavitt, I.; Ewing, D. W.; Del Bene, J. E. *J. Am. Chem. Soc.* **1991**, *113*, 9389.
5. <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 0.25 (s, 18H), 1.49 (s, 2H); <sup>13</sup>C NMR: -0.89, 28.45, 43.95, and 66.42 ppm. Anal. Calcd. for C<sub>11</sub>H<sub>20</sub>Cl<sub>4</sub>Si<sub>2</sub>: C, 37.72; H, 5.76; Cl, 40.49. Found: C, 37.90; H, 5.48; Cl, 40.28.
6. Seyferth D. *Acc. Chem. Res.* **1972**, *5*, 65.
7. Frisch, M. J.; *et. al.*, Gaussian 99, Development Version (Revision 0.2), Gaussian, Inc., Pittsburgh, PA, 1998.